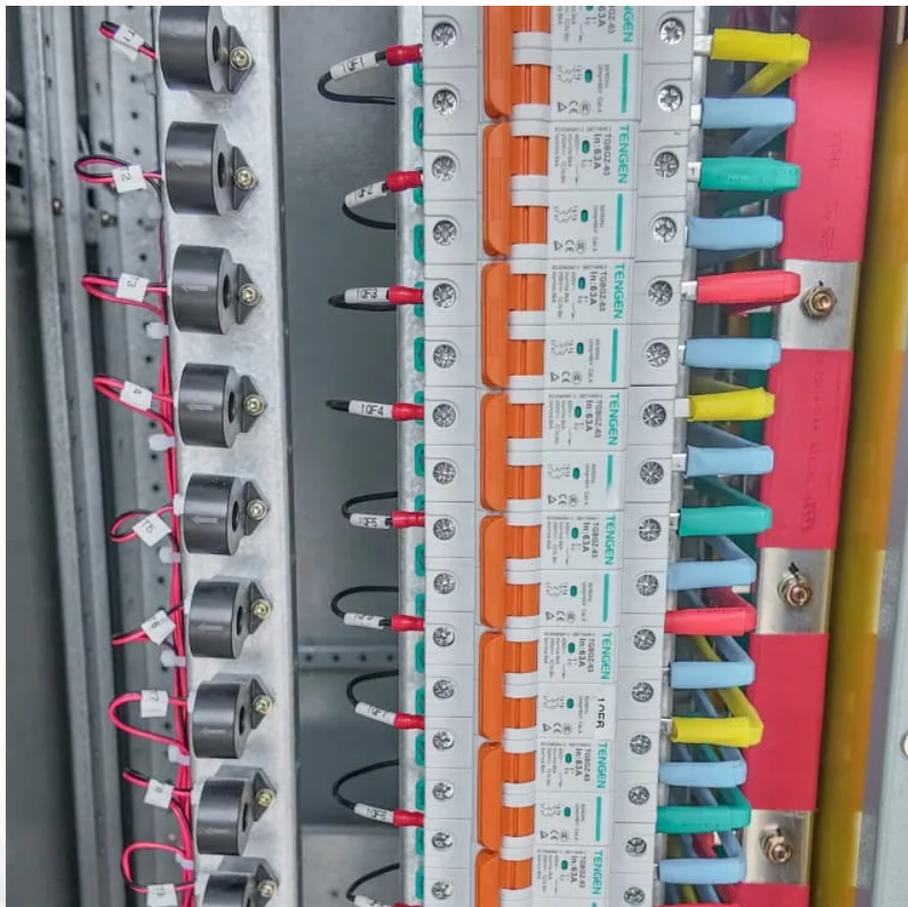


# What are the functions of aluminum sheets for energy storage batteries





## Overview

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Why is aluminum used in batteries?

Historically, aluminum has been employed in batteries primarily as a casing material or a current collector due to its lightweight and conductive properties. These roles, while important, position aluminum as a passive component within the battery architecture.

Can aluminum batteries be used as rechargeable energy storage?

Secondly, the potential of aluminum (Al) batteries as rechargeable energy storage is underscored by their notable volumetric capacity attributed to its high density ( $2.7 \text{ g cm}^{-3}$  at  $25 \text{ }^\circ\text{C}$ ) and its capacity to exchange three electrons, surpasses that of Li, Na, K, Mg, Ca, and Zn.

Are aluminum-ion batteries the future of energy storage?

Aluminum-ion batteries exhibit impressive performance metrics that position them as a viable competitor to lithium-ion systems. Key performance indicators such as energy density, cycle life, and charging time highlight the potential of aluminum-based technology to revolutionize the energy storage landscape.

What is an aluminum battery?

In some instances, the entire battery system is colloquially referred to as an “aluminum battery,” even when aluminum is not directly involved in the charge transfer process. For example, Zhang and colleagues introduced a dual-ion battery that featured an aluminum anode and a graphite cathode.



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